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practical part of the course covers experiments on the synthesis of organic compounds and drugs using different analysis and purification techniques and includes synthesis of chiral drugs and the techniques used in the separation of enantiomers.

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depth understanding
of organic chemistry
is core to the study of
this subject as it plays
a key role in many of
the processes utilised
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PROCEDURE:
STEP-1: In a 2 lit RBF
fitted with a double

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surface reflux

condenser place,

11.5gm (0.5gm atom)

of clean sodium and it

is dissolved in 250ml

of absolute alcohol.

To this solution add

80gm (0.5mole of

ethyl malonate)

followed by 30gm

(0.5mole of dry urea)

dissolved in 250ml of

ethanol.

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the laboratory

activities to

understand the model

of ionic compound

close packing,

reduction-oxidation

reaction of several

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metals, improve the double salt preparation and re-crystallization skills as the basic for advance laboratory activities.

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The units of volume commonly used in this manual are cm^3 and

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dm³. The conversion between cm³ and dm³ is 1 dm³ = 1000 cm³. Note that a cm³ are the same as millilitres (ml) and dm³ are the same as litres (L). 1 cm³ = 1 ml and 1 dm³ = 1 L. B.

Concentration

Concentration is a measure of amount per volume. The amount can be

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grams, kilograms,

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and proteins in

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by Charles Dickson

Pharmaceutical chemistry seeks to examine the synthetic methodology used in order to produce pure biologically active molecules to be formulated as medicines. An in depth understanding of organic chemistry is core to the study of this subject as it plays

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a key role in many of
the processes utilised
in the formation of
medicines.

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...

1) Collect the following items which you will need to carry out the experiment.

- (a) 2 100mL round bottom flask
- (b) 1 spatula
- (c) 1 cork ring
- (d) 1 plastic bowl
- (e) 1

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ceramic Buchner
funnel (f) 1 watch
glass (g) 1 magnetic
stirrer bar (h) 1 zip
lock bag. 2) Put on a
pair of gloves before
you start the
experiment.

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...

Step 1: First step

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involves the formation of Acetylium ion, the reaction is enhanced by proton donation by concentrated acid.

Step 2: Amino group of Para-Aminophenol dissociates as follows:

Step 3: Acetylium ion attacks the anion at the nitrogen atom to form Para-acetyl Aminophenol or Paracetamol.

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introduction this
course introduces the
basic concepts of
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