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assume that you  
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 $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

Determine the  
number of grams in  
each of the quantities  
below. 1, 2.5 moles of  
 $\text{NaCl}$  2. 0.50 moles of  
3, 1.70 moles of  
 $\text{KMnO}_4$  4, 0.25 moles  
of  $\text{KCl}$  ...

Molarity Chemistry

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Answer Key 1. Zinc  
and Lead (II) nitrate  
react to form Zinc

Nitrate and Lead.  $Zn$   
+  $Pb(NO_3)_2$   $Zn(NO_3)_2$  +  $Pb$  Single

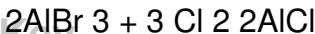
Replacement 2.

Aluminum Bromide  
and Chlorine gas  
react to form

Aluminum Chloride

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and Bromine gas.



+ 3Br<sub>2</sub> Single

Replacement 3.

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VOLUME-VOLUME

PROBLEMS I. N2

Name What volume of

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hydrogen is necessary to react with five liters of nitrogen to produce ammonia? (Assume constant temperature and pressure.) 5 L 3  
"Ids I-I 2. What volume of ammonia is produced in the reaction in Problem 1?

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The Gas Laws  
CHARLES'S LAW

Charles' Law states  
the volume of a gas  
varies directly with the  
Kelvin temperature,  
assuming the  
pressure is constant.

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And a variation of the law states that pressure of a gas varies directly with the Kelvin temperature, ...  
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