

Acid Nomenclature Answer Key

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Acid Nomenclature Worksheet KEY Write the formula for each of the acids listed below: 1. Nitric acid HNO₃ 2. Chloric acid HClO₃ 3. Acetic acid HC₂H₃O₂ 4. Hydrobromic acid HBr 5. Sulfurous acid H₂SO₃ 6. Chlorous acid HClO₂ 7. Hydrochloric acid HCl 8. Phosphoric acid H₃PO₄ 9. Nitrous acid HNO₂ 10. Hydrofluoric acid HF 11. Perchloric acid HClO₄ 12. Hydroiodic acid HI 13.

[Acid Nomenclature Worksheet Name](#)

HCl = hydro chlor ic acid = hydrochloric acid HBr = hydro brom ic acid = hydrobromic acid HF = hydro fluor ic acid = hydrofluoric acid
2. NAMING OXYACIDS: These are more difficult to name because these acids have hydrogen, a nonmetal, and may have varying numbers of oxygen atoms.

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ANSWER KEY Name the following compounds or draw structures for the names. 1. CH₃CH₂CO₂H propanoic acid . 2. ethyl butyl ether CH₃-CH₂-O-CH₂-CH₂-CH₂-CH₃ ... 150 People Used View all course >>. Visit Site. Nomenclature

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Practice Quiz. Answers are on the last page ...

~~Nomenclature Practice Answer Key—12/2020~~

Key Points. Acids are named based on their anion — the ion attached to the hydrogen. In simple binary acids, one ion is attached to hydrogen. Names for such acids consist of the prefix "hydro-", the first syllable of the anion, and the suffix "-ic". Complex acid compounds have oxygen in them.

~~Naming Acids and Bases | Introduction to Chemistry~~

Acids are an important class of compounds containing hydrogen and having special nomenclature rules. Binary acids are named using the prefix hydro-, changing the -ide suffix to -ic, and adding "acid;" HCl is hydrochloric acid. Oxyacids are named by changing the ending of the anion to -ic, and adding "acid;" H₂CO₃ is carbonic acid.

~~2.7 Chemical Nomenclature—Chemistry~~

Acid Nomenclature Worksheet KEY Write the formula for each of the acids listed below: 1. Nitric acid HNO₃ 2. Chloric acid HClO₃ 3. Acetic acid HC₂H₃O₂ 4.

~~Nomenclature Worksheet 2 Answer Key~~

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Read PDF Naming Acids And Bases Answer Key to show that. $H^+ (aq) + OH^- (aq) \rightarrow H_2O(l)$ is an acid-base reaction in the Lewis sense as well as in the Arrhenius and Brønsted-Lowry senses. 10.E: Acids and Bases (Exercises) - Chemistry LibreTexts i memorized the strong and weak acids and bases, but if

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~~Acids And Bases Answer Key — Teacher Worksheets~~

acid: HI (aq) hydroiodic acid: CH_3COOH acetic acid: HF (aq) hydrofluoric acid: H_3PO_3 phosphorous acid: HCN (aq) hydrocyanic acid: HClO_3 chloric acid: H_2CO_3 carbonic acid: H_2SO_3 sulfurous acid: HClO_2 chlorous acid: HNO_3 nitric acid: HBr (aq) hydrobromic acid

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